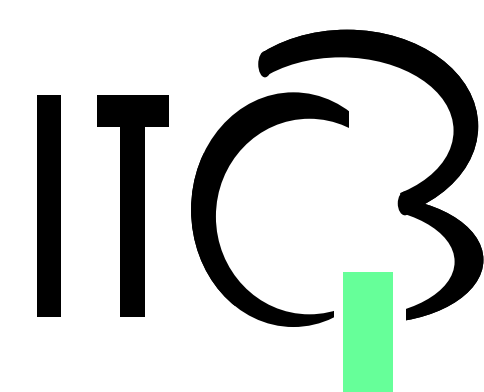


Mo(VI) Catalysts Obtained from η^3 -Allyl Dicarbonyl Precursors: Synthesis, Characterization and Catalytic Performance in Cyclooctene Epoxidation

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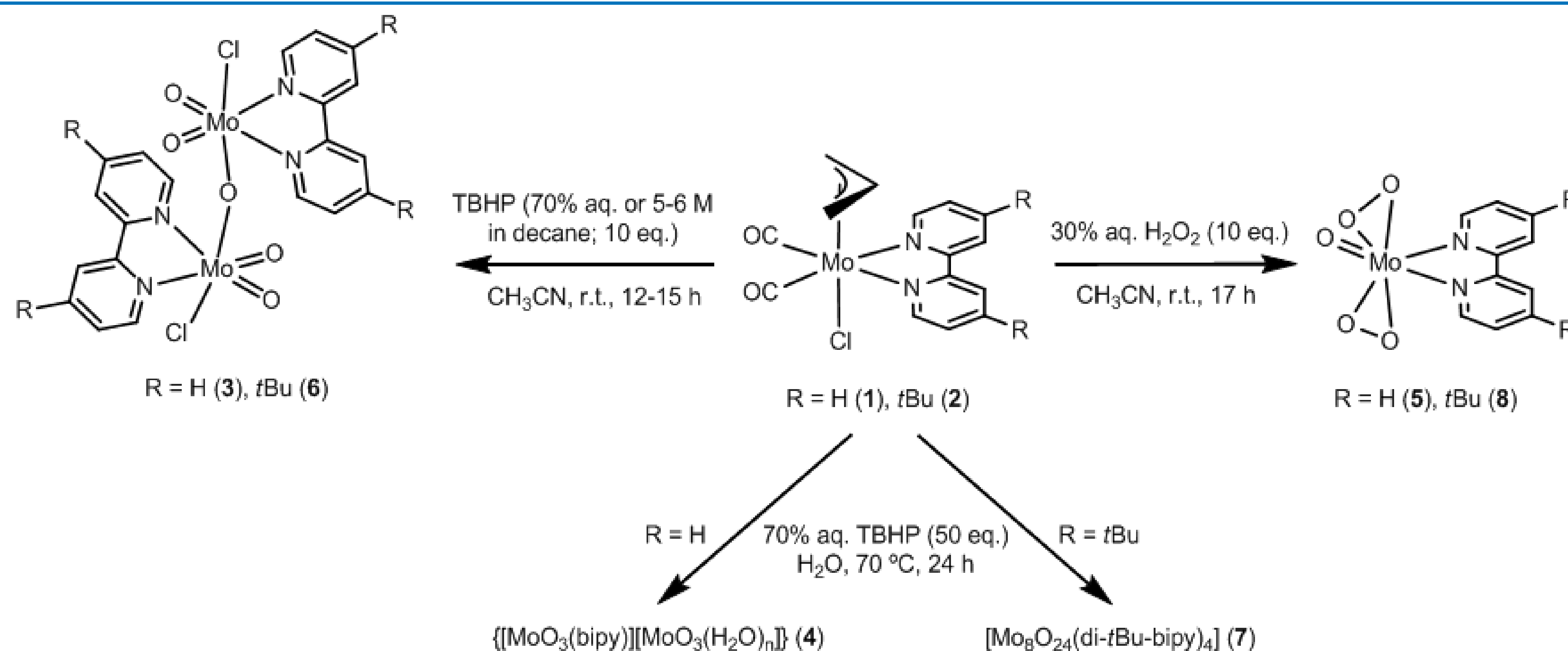
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Introduction

Allyl dicarbonyl complexes $[\text{Mo}(\eta^3\text{-C}_3\text{H}_5)\text{X}(\text{CO})_2(\text{L})_n]$ (X = halide) have been found to act as catalyst precursors for several reactions, namely the polymerization of dienes and the epoxidation of olefins.¹ In our recent investigations of complexes $\text{cis-}[\text{Mo}(\text{CO})_4(\text{L})]$, we found that the nature of the bi-dentate ligand L influences the species formed by oxidative decarbonylation and their catalytic performance in olefin epoxidation: with L = bipy, the organic-inorganic hybrid $[\text{MoO}_3(\text{bipy})]$ is obtained; with L = di-*t*Bu-bipy, the polynuclear complex $[\text{Mo}_8\text{O}_{24}(\text{di-}t\text{Bu-bipy})_4]$ is obtained instead.² Here we report on the use of the complexes $[\text{Mo}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}(\text{CO})_2(\text{L})]$ (L = bipy (**1**), di-*t*Bu-bipy (**2**)) as catalyst precursors in the epoxidation of cyclooctene (Cy) using aqueous TBHP or H_2O_2 as oxidant. Additionally, oxidative decarbonylation was performed for both catalysts, in aqueous media, and the obtained products were characterized and also used as catalysts.³

Synthesis



Characterization

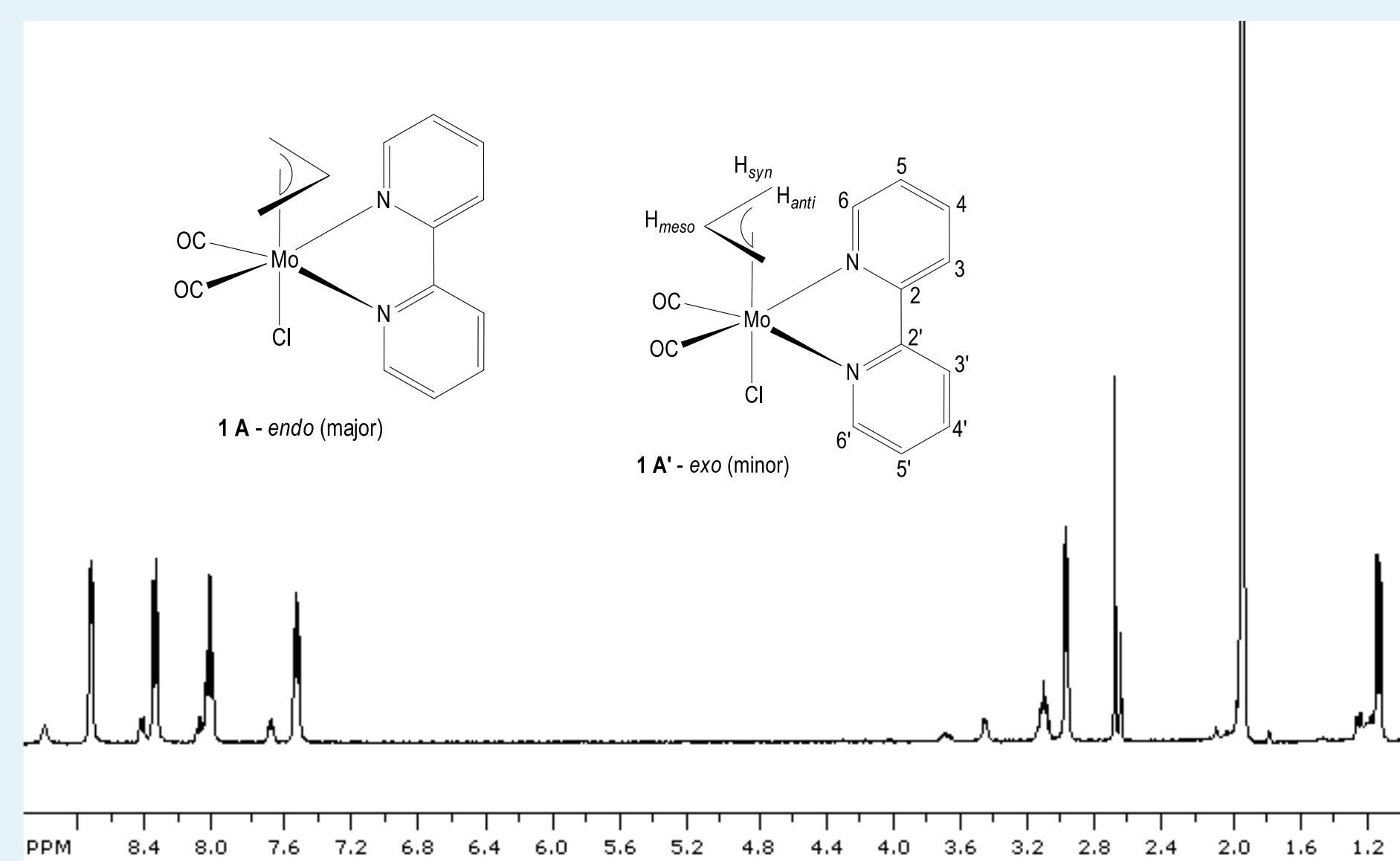


Figure 1. ^1H NMR spectrum of **1** in acetone- d_6 at r.t.

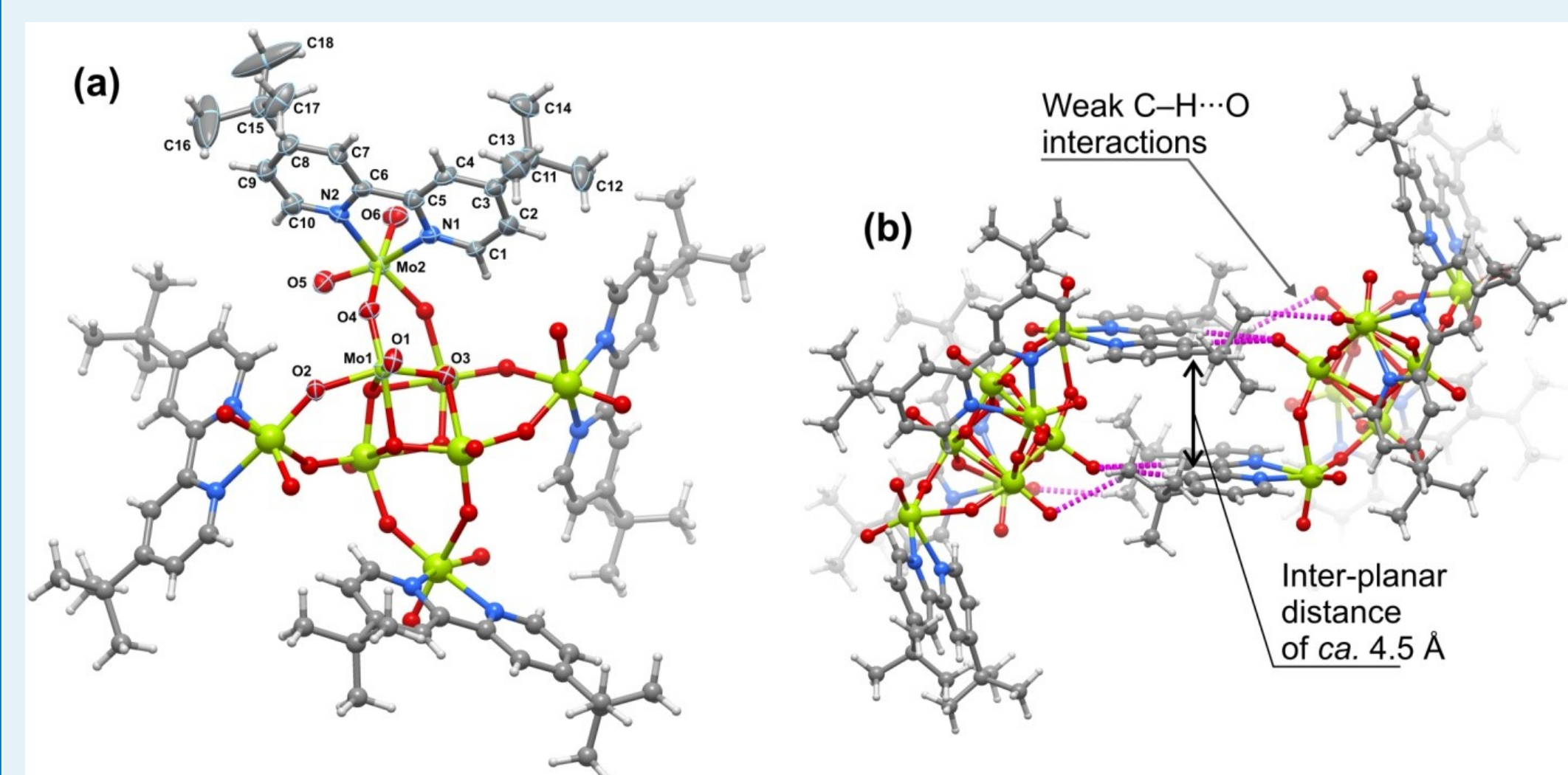


Figure 2 (a) Schematic representation of the windmill-type complex $[\text{Mo}_8\text{O}_{24}(\text{di-}t\text{Bu-bipy})_4]$ (**7**). Thermal ellipsoids drawn at the 50% probability level. (b) Portion of the crystal structure of **7** depicting the most structurally robust $\text{C-H}\cdots\text{O}$ short contacts.

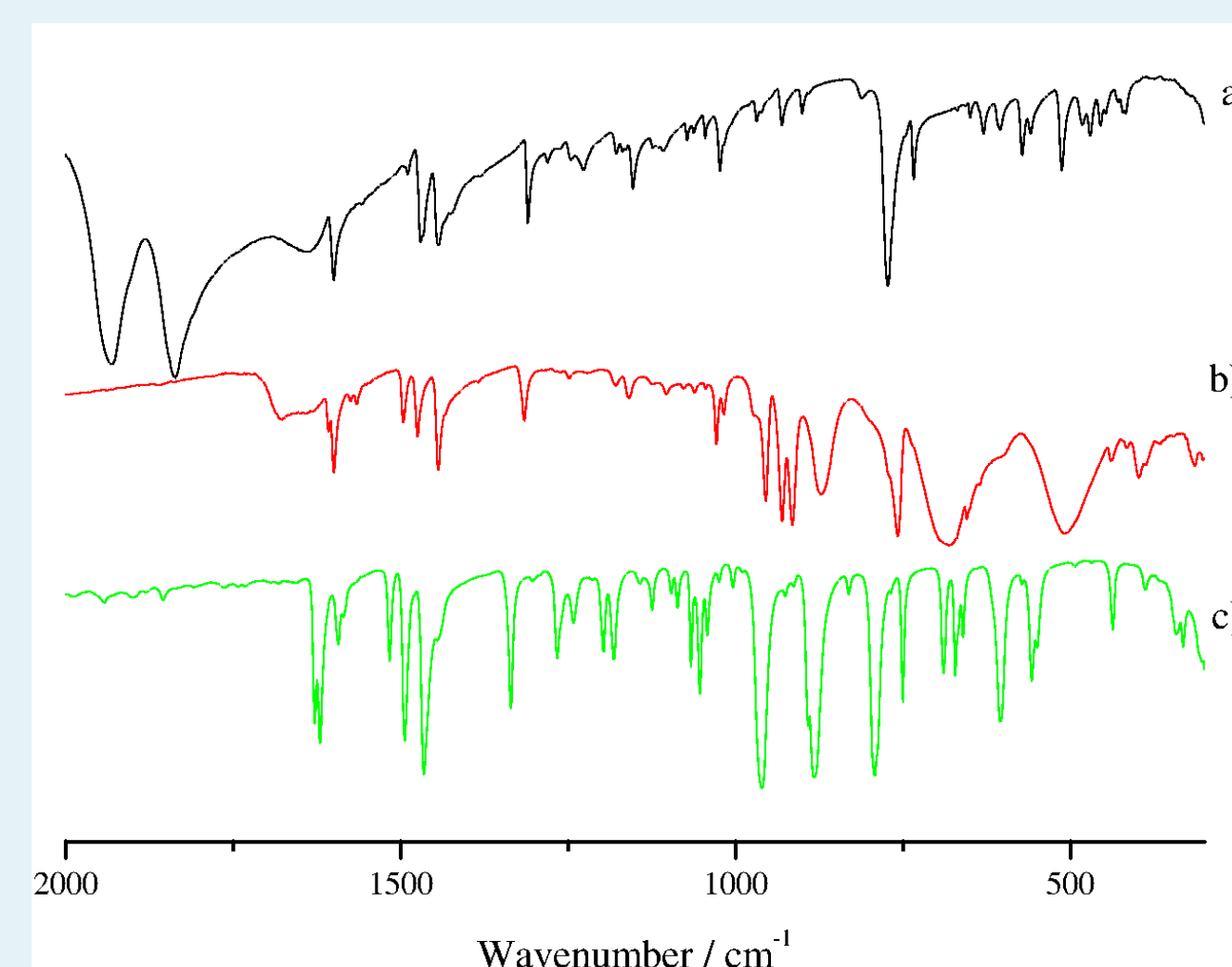


Figure 3. FT-IR spectra of compounds **1** (a), **4** (b), **5** (c)

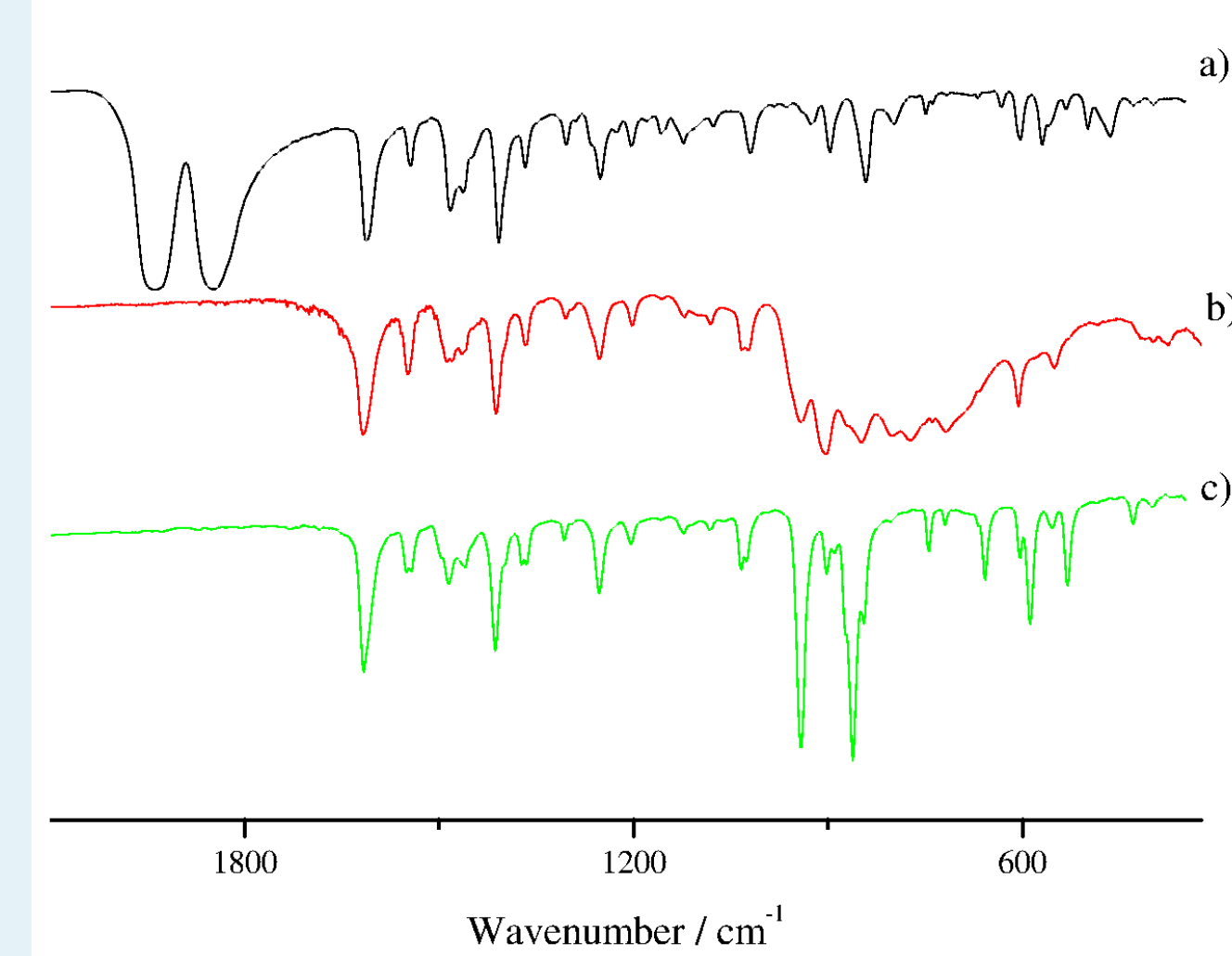


Figure 4. FT-IR spectra of compounds **2** (a), **7** (b), **8** (c)

Comp.	Selected FT-IR (cm^{-1})
1	1932vs, 1837vs (CO), 1600m (C=N bipy)
3	1600s (C=N bipy), 931vs, 903vs (Mo=O), 789vs (Mo–O–Mo)
4	955s, 930s, 915s, 868s (Mo=O), 682s (Mo–O–Mo), 514s (OMo ₃)
5	939vs (Mo=O), 861s (O–O), 651m, 583m, 536m (Mo(O ₂) ₂)
7	942 vs, 903vs (Mo=O), 849s, 773s, 719s (Mo–O–Mo)

With the exception of **3**, the oxo Mo(VI) compounds have been previously prepared by different routes: **4**,^{4a} **5**,^{4b} **6**,^{4c} **7**,² and **8**.^{4d}

Catalysis

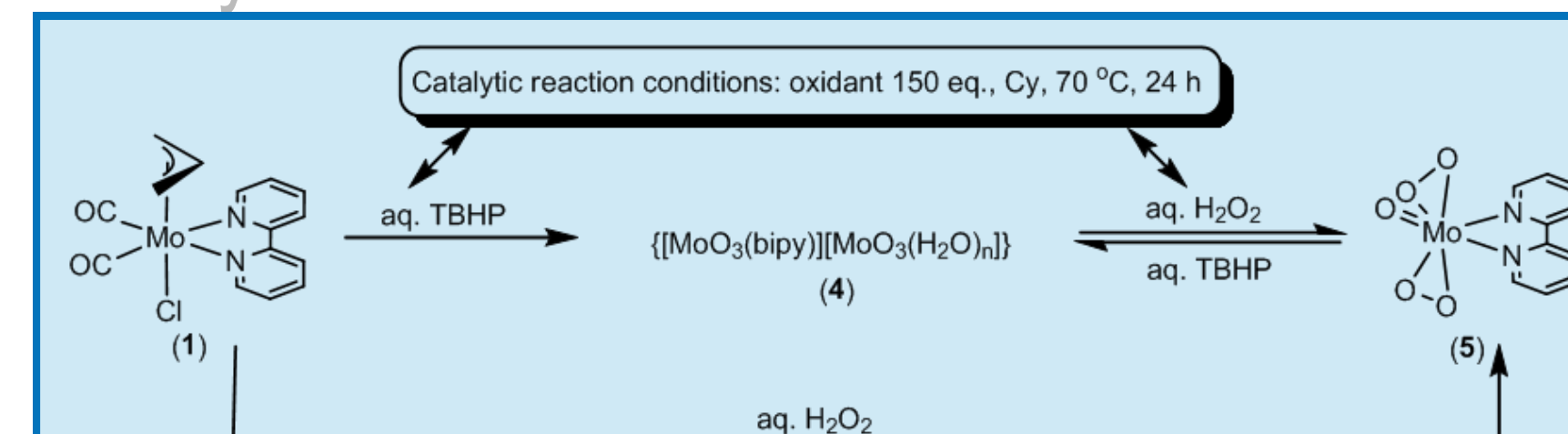


Table 1. Catalytic epoxidation of Cy with TBHP^a

Compound	Solvent	Yield (%) ^b	Identified solid after 24h ^c
1	H_2O	37	4 (TY≈60%)
1	CH_3CN	19	4
3	H_2O	32	4
4	H_2O	38	4
5	H_2O	28	4
2	H_2O	87	—
2	CH_3CN	59	—
6	H_2O	96	—
7	H_2O	90	—
8	H_2O	98	—

Table 2. Catalytic epoxidation of Cy with H_2O_2 ^a

Compound	Solvent	Yield (%) ^b	Identified solid after 24h ^c
1	CH_3CN	66	5
3	CH_3CN	71	5
4	CH_3CN	79	5
5	CH_3CN	54	5
2	CH_3CN	26	8
6	CH_3CN	54	8
7	CH_3CN	75	8
8	CH_3CN	81	8

^a mmol of $\text{Mo}:\text{Cy}:\text{oxidant} = 0.018:1.8:2.75$, 24 h reaction, 70 °C, 800 rpm, total volume ≈ 1.25 mL; ^b CyO yield (100% selectivity); ^c solid formed identified by FT-IR spectroscopy; TY = theoretical wt.% yield of product formed.

Conclusions

The η^3 -allyl dicarbonyl complexes $[\text{Mo}(\eta^3\text{-C}_3\text{H}_5)\text{Cl}(\text{CO})_2(\text{L})]$ (L = bipy, **1**; di-*t*Bu-bipy, **2**) are convenient precursors to oxo Mo(VI) compounds that selectively catalyze the epoxidation of Cy in aqueous media. Reaction of **1** or **2** with the oxidant results in oxidative decarbonylation. When the oxidant is H_2O_2 aq., the oxodiperoxo complexes $\text{MoO}(\text{O}_2)_2(\text{L})$ (**5**, **8**) are formed. By contrast, when the oxidant is TBHP aq., different oxo Mo(VI) compounds are formed, depending on the ligand L and the reaction conditions: oxo-bridged dimers $[\text{MoO}_2\text{Cl}(\text{L})]_2\text{O}$ (**3**, **6**), the hybrid $\{[\text{MoO}_3(\text{bipy})][\text{MoO}_3(\text{H}_2\text{O})_n]\}$ (**4**), the octanuclear complex $[\text{Mo}_8\text{O}_{24}(\text{di-}t\text{Bu-bipy})_4]$ (**7**).

We anticipate that the extension of these studies to other η^3 -allyl dicarbonyl complexes, including chiral derivatives, will provide novel oxo Mo(VI) compounds, possibly active in aqueous catalysis.

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Acknowledgments

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